

Science Curriculum Map Overview Chemistry

Topic	Skills	Approximate Weeks of Study
Module 1 Structure and Properties of Matter	<ul style="list-style-type: none"> → Use the periodic table as a model to predict the relative properties of elements based on the patterns of electrons in the outermost energy level of atoms. [Periodic Table] → Plan and conduct an investigation to gather evidence to compare the structure of substances at the bulk scale to infer the strength of electrical forces between particles. [Intermolecular forces] → Develop models to illustrate the changes in the composition of the nucleus of the atom and the energy released during the processes of fission, fusion, and radioactive decay. [Nuclear Chemistry] → Communicate scientific and technical information about why the particulate-level structure is important in the functioning of designed materials [states of matter] → Analyze data to support the claim that the combined gas law describes the relationships among volume, pressure, and temperature for a sample of an ideal gas. [Gases] → Use evidence to support claims regarding the formation, properties and behaviors of solutions at bulk scales. [Solutions] 	12 Weeks
Module 2 Chemical Reactions	<ul style="list-style-type: none"> → Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties. [Bonding] → Develop a model to illustrate that the release or absorption of energy from a chemical reaction system depends upon the changes in total bond energy. [Kinetics] → Apply scientific principles and evidence to explain how the rate of a physical or chemical change is affected when conditions are varied. [Collision theory] → Refine the design of a chemical system by specifying a change in conditions that would produce increased amounts of products at equilibrium. [Equilibrium] → Use mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction. [Balancing] → Plan and conduct an investigation to compare properties and behaviors of acids and bases. [Acid/Base] → Use evidence to illustrate that some chemical reactions involve the transfer of electrons as an energy conversion occurs within a system. [Oxidation/Reduction] 	26 Weeks